**Phytochemical Screening and Investigation of Antimicrobial Activity from Derivatives of Aveloz (Euphorbia tirucalli L.)**

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**Abstract**

Aveloz (Euphorbia tirucalli L.) is a plant known for its therapeutic activities and the toxicity of latex. Antimicrobial, antitumoral, anti-inflammatory, anti-ulcerative activities, among others, are examples of its medicinal properties already described in different studies. On the other hand, extrinsic and intrinsic factors associated with the plant metabolism are able to produce different metabolites among samples of the same species. Therefore, studies of antimicrobial activity and the phytochemical screening of different specimens of this plant are relevant, since it is able to aggregate knowledge about its properties widely used by the population. In the present study, the phytochemical screening was performed and the antimicrobial activity of four extracts from aerial parts was investigated. The results showed that the material presents antimicrobial and anti-inflammatory potencies, since the phytochemical screening revealed the presence of flavonoids and, through the susceptibility tests carried out on 96-wells microplates with viability stains, it was verified the occurrence of growth inhibition of Pseudomonas aeruginosa and Escherichia coli strains. The plant derivatives presenting significant antimicrobial activity were the ethyl acetate and the butanol extracts. Comparing these data to the literature we conclude that new researches using different specimens of E. tirucalli may provide findings with potential application in medicine.

**Keywords:** Euphorbia tirucalli L., aveloz, phytochemical screening, antimicrobial potential.

**Introduction**

Historically, medicinal plants have always been an important source of compounds for drugs production, in the form of pure active principles or as traditional preparations. Therefore, using natural resources for therapeutic purposes is considered an ancient practice and can be equated with the emergence of human civilization [1,2].

Currently, several factors support the use of medicine plants, such as the wide range of adverse effects of chronic use of industrialized drugs with chronic use, increased consumption of natural products by the population, and a tendency to use the traditional medicine based on the concept that it could be free of adverse effects. However, it is known that it does not present scientific support, but it has been passed down through the generations [3].

Self-medication, probably originated in traditional and/ or family practices trough years, leads to the indiscriminate use not only of plant drugs in home preparations, but also of industrialized drugs. This fact leads to drugs interactions, which can lead to severe reactions such as the potentiation of the drug effect or to the inactivation of the active substances. In addition, the toxic effects of many plant drugs are still unknown and represent a potential risk to the population [3].

In recent decades, the Brazilian government has been interested in linking popular knowledge about medicinal plants to the sustainable development, and the Resolution RDC No. 17, of February 24, 2000, was published, classifying phytotherapeutics as medicines [4].

Euphorbia tirucalli Lineau, popularly known as aveloz, has a high toxicity in its latex (part of the plant most commonly used). This plant belongs to the family Euphorbiaceae and originates from Africa and America. Later it was introduced in Brazil and has adapted to the hot climate regions of the country, especially in the North and Northeast, where it has propagated as an ornamental plant widely used as protection against invaders [5].

Morphologically, this plant can be described as presenting extremely cylindrical green and branched stem with small leaves.
that appear only at the beginning of its development, small female and male terminal flowers and encapsulated fruit containing three ovoid seeds (Figure 1).

The stem of *E. tirucalli* produces caustic and toxic latex (Figure 1C), that has been used to treat some diseases, such as helminthiasis, syphilis, and tumors. This latex is extremely dangerous for domestic use, since it is corrosive in contact with the skin, which can cause lesions, itching, edema, burns and even tissue necrosis. In contact with the eyes it causes swelling, burning, blurred vision and, if the victim does not have prompt medical attention, the lesions may progress to the cornea’s destruction causing permanent blindness. When ingested, latex may cause vomiting, diarrhea, and bleeding due to the irritation of the gastric mucosa. Ingestion of high doses of latex diluted in water can cause vasoconstriction, tissue hypoxia and increase of blood clotting. Due to these characteristics, it was cataloged in the National Information Program on Toxic Plants of the Oswaldo Cruz Foundation [7-11].

Additionally, in Brazil, there are reports of the use of *E. tirucalli* latex by the population as an antimicrobial agent, besides being used as a treatment of intestinal constipation, asthma, coughs, otitis, arthralgias and papillomas. However, the dosage used for these different types of pathologies varies based only on the empirical knowledge [12-15].

In this sense, *E. tirucalli* is one of the plant species exploited by the scientific community. And, in order to contribute to increase the knowledge about the use of this plant, the present study evaluated the potential antimicrobial activity of the extracts obtained from the aerial parts of *E. tirucalli* and carried out a phytochemical screening to identify the classes of chemical substances present in these plant extracts.

**Material and Methods**

**Collection of the plant drug**

The aerial parts of the plant drug *E. tirucalli*, popularly known in Brazil as graveto-do-cão, figueira-do-diabo, dedo-do-diabo, pau-pelado, São Sebastião’s tree, aveloz or avelós, were collected from the Piatã garden at the Faculty of Medical Sciences and Health of Juiz de Fora (FCMS / SUPREMA - JF) on 05/31/2017. Its geographical location is given by the following coordinates -21.8163711 / -43.3816181. The plant was identified by the curator Dr. Luiz Menini Neto, and the sample of material was registered in the Leopoldo Krieger Herbarium - CESJ 19803, at the Federal University of Juiz de Fora - Minas Gerais - Brazil.

**Preparation of plant derivatives**

The collected samples were hand-cut and cut with scissors, which sizes ranging from 2.0 to 4.0 cm and were later weighed with a scale in a beaker: 91.7g of the sample were placed in a glass container then, in the chapel of exhaustion, 917mL of hexane (C₆H₁₄) were added. After seven days, the supernatant was collected, duly stored and identified in an amber glass vial. In the glass container with the sample, 917mL of dichloromethane...
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(C₇H₆Cl₃) were added and again, after seven days, that supernatant was collected, duly stored and identified in an amber glass vial. In the glass container with the sample, 917mL ethyl acetate (CH₃COOCH₂CH₂) were added, and the same method was followed with the supernatant obtained, also repeating it with butanol (C₄H₈O₂). For the addition of the solvents and transposition of the supernatants, a 250 mL graduated beaker (3 complete dosages plus a dosage of 167 mL) was used. The containers were covered with aluminum foil with the shiny part turned inwards and, the extract was covered with a clear PVC film. Finally, the extracts were concentrated by evaporation of the solvents in a rotary evaporator, resulting in: hexane extract (EHEX); dichloromethane extract (EDCM); ethyl extract (EACET) and butanol extract (EBUT).

Evaluation of the antimicrobial activity of plant derivatives

In order to execute the susceptibility tests with the plant derivatives, cultures of the following microorganisms were used: Staphylococcus aureus (ATCC 6538); Pseudomonas aeruginosa (ATCC 15442) and Escherichia coli (ATCC 25922) in liquid culture (Brain Heart Infusion - BHI). The tests were carried out in the log phase of bacterial development.

96-well microtiter plates were used to verify the antimicrobial potential of the plant derivatives, using resazurin solution as a bacterial viability developer. The solutions tested were initially solubilized in DMSO (Dimethyl Sulfoxide) and then in the liquid culture (BHI). The final concentration of DMSO in these solutions was 10%. The wells of the margins of each microplate were filled with sterile water in order to reduce the evaporation of the samples, which were occupying the inner part of the microtiter plates. The water-free wells were filled with 100μL of BHI culture. Then 100μL of the solutions of the plant derivatives to be tested were added to the first wells of the “B” line in the microtiter plates. From these, serial dilutions were performed for the wells of the following lines (B to C, C to D...). Then 100μL of the bacterial suspensions (10⁵CFU / mL) were added and the final concentrations of the solutions of the tested materials in the microtiter plates were: EHEX - 375 μg/mL, 187.5 μg/mL, 93.75 μg/mL, 46.875 μg/mL; EDCM - 625 μg/mL, 312.5 μg/mL, 156.25 μg/mL, 78.125 μg/mL; EACET - 250 μg/mL, 125 μg/mL, 62.5 μg/mL; EBUT - 500 μg/mL, 250 μg/mL, 125 μg/mL, 62.5 μg/mL. Ciprofloxacin was used as a positive control, for the reaction with ferric chloride solution containing 2.0 mL of 4.5% ferric chloride solution was added to tube 2 and for reaction with sodium hydroxide, 1.0 mL of 5% sodium hydroxide solution was added to the tube 3.

For the reaction with aluminum chloride, a circular cut of filter paper was used and 2 drops of the hydroalcoholic extract were added in different places. 50 μL of the aluminum chloride solution was added to one of them. The paper was then observed under ultraviolet light.

Alkaloids: 20.0 mL of 1% HCl were added to 2.0 g of the plant drug. The solution was heated without boiling for 2 minutes and, at room temperature, it was filtered into a separatory funnel. Thereafter, 10% ammonium hydroxide was added until the mixture became alkaline, which was observed with a pH in an indicator paper. To this solution it was added 5.0 mL of chloroform, and then the aqueous and organic layers were separated and collected. The organic extract was concentrated and 2.0 mL of 1% HCl was added thereto. 50.0 μL of the acid solution obtained and 50.0 μL of a general alkaloid reaction were placed in 6 test tubes (Dragendorff, Mayer, Bertrand, Bouchardat, Silicotungstic Acid and Picric Acid, respectively). The solution and the reactions were mixed with glass rods and the reactions were observed.

Cardiotonics: 20.0mL of 50% ethanol was added to 2.0 g of the plant drug. This solution was boiled and then filtered at room temperature. Then 5.0 mL saturated lead acetate solution was added. This solution was centrifuged and the supernatant was placed on a separatory funnel. To this solution was added 5.0 mL of chloroform and the aqueous and organic layers were separated and collected. The chloroform extracts obtained were distributed in 3 test tubes (tubes 1, 2 and 3) and concentrated in a water bath. In tube 1, the Liebermann-Bouchard reaction was carried out. The residue was taken up with 0.5 mL of acetic anhydride and, by the walls of the test tube; 1.0 mL of concentrated sulfuric acid was carefully added, without stirring. In tube 2, the Kedde reaction was performed. To the residue were added 250 μL of 1% alcoholic solution of 3,5-dinitrobenzoic acid 100 μL of 1N KOH solution. In tube 3, the Keller-Killiani reaction occurred. The residue was dissolved in about 1 mL of glacial acetic acid and then 100 μL of 2% ferric chloride was added. Then, carefully, by the walls of the tube, 1.0 mL of concentrated sulfuric acid was added, without stirring.

Phytochemical screening

The phytochemical analysis consisted on the detection of the following chemical classes of metabolites derived from secondary metabolism [16].

Flavonoids: 20 mL of 70% ethanol were added to 5.0g of the plant drug and heated until boiling. The hydro-alcohol extract was filtered at room temperature and distributed into 4 tubes containing 3.0 mL each. One of the tubes was used as negative control (tube 4) and the others (tubes 1, 2 and 3) were used in the identification reactions.

For the Shinoda reaction, 1 metal magnesium fragment and 1.0 mL concentrated hydrochloric acid was added to tube 1; for the reaction with ferric chloride, 1.0 mL of the 4.5% ferric chloride solution was added to tube 2 and for reaction with sodium hydroxide, 1.0 mL of 5% sodium hydroxide solution was added to the tube 3.

Saponins: 10.0 mL of distilled water was added to 1.0 g of the drug. The solution was boiled and, after cooling, the obtained extract was vortexed for 15 seconds. It was observed if there was any foam formation in it.

Tannins: 10.0 g of the plant drug was diluted in 100.0 mL of distilled water. The solution was boiled for 3 minutes, cooled and filtered. The filtrate was distributed into 5 test tubes (1, 2, 3, 4, and 5) and the identification reactions were assayed. Tube 1 was used as negative control. For the reaction with gelatin, 150 μL of 2% gelatin solution were added to tube 2. For the reaction with alkaloids, 150 μL of 1% quinine hydrochloride were added to tube 3. For heavy metal reactions 150 μL of 4% Cu(ACO)₂ solution were added to tube 4, and in tube 5 were added 150 μL of the 10% Pb(ACO)₂ solution.

Anthraquinones: In 3.0 g of the drug was added 1.00 mL of ether. The extract formed was collected and put in a test tube. The procedure was repeated twice, collecting the extracts obtained and putting it in the same test tube. To this was added 1.0 mL of 10% ammonium hydroxide. The mixture was stirred for about 1.0 minute. It was observed if there was the presence of free anthraquinones.

Investigation of O-Glycosides was performed as follows: to the reserved plant drug of the first experiment, 20.0 mL of distilled water was added and boiled for 2 minutes. Then, 5.0 mL of 10% HCl was added and boiled for another 2 minutes. The mixture was filtered to a separatory funnel and 8.0 mL of hexane was added. The organic layer was collected and the aqueous acid layer was reserved for the next experiment. 3mL of diluted ammonia was added to the organic layer, stirred, and then allowed to stand.

For the identification of C-glycosides, 3.0 mL of 5% ferric chloride were added to the acidic aqueous phase and boiled for 3 minutes. The mixture was filtered to a separatory funnel and to it was added 8.0 mL of hexane. Thereafter, the organic layer was collected in a test tube and the acidic aqueous layer was discarded. To the organic layer 3.0 mL of ammonia was added. The mixture was stirred, and then it was allowed to stand.

Results and Discussion

Antimicrobial susceptibility tests revealed the occurrence of antimicrobial activity for almost all the extracts against at least one of the bacterial species tested. The most active plant derivatives were ethyl acetate and butanolic extracts, however the S. aureus species’ growth wasn’t inhibited by any of the extracts tested. Vale et al. (2011), using aerial parts of aveloz, identified the absence of antimicrobial of activity dry extract on S. aureus and detected an inhibitory effect using the fresh extract [17]. Growth inhibition of P. aeruginosa and E. coli was identified in the present study, as well as in studies by Upadhyay et al. (2010) [14].

In the present study, latex did not have its antimicrobial potential evaluated; however previous studies have shown absence of bactericidal activity in this plant derivative [18-20].

The results for the tests using the reference antibiotic (ciprofloxacin) showed that the microorganisms tested were sensitive to it as shown in other studies (Table 1) [21,22].

<table>
<thead>
<tr>
<th>Materials Tested</th>
<th>P. aeruginosa CIM (µg/mL)</th>
<th>E. coli CIM (µg/mL)</th>
<th>S. aureus CIM (µg/mL)</th>
</tr>
</thead>
<tbody>
<tr>
<td>EHEX</td>
<td>187.5</td>
<td>n/a</td>
<td>n/a</td>
</tr>
<tr>
<td>EDCM</td>
<td>n/a</td>
<td>250</td>
<td>n/a</td>
</tr>
<tr>
<td>EACET</td>
<td>62.5</td>
<td>250</td>
<td>n/a</td>
</tr>
<tr>
<td>EBUT</td>
<td>62.5</td>
<td>250</td>
<td>n/a</td>
</tr>
<tr>
<td>Ciprofloxacin</td>
<td>3.125</td>
<td>0.37</td>
<td>6.25</td>
</tr>
</tbody>
</table>

Despite the antimicrobial activity identified in the present study, the toxic potential of this plant acts as a limiting characteristic of its use as a source of antimicrobrial potential. In this sense, tests performed by Silva et al. (2007) on the toxicity of aqueous extract of aveloz in pregnant rats did not identify clinical signs of toxicity [5]. On the other hand, Waczuk (2014) and Neodini (2015) found toxic activity in the aqueous and ethanolic extracts of the aerial parts of the aveloz [23,24]. In relation to the latter, the mechanisms of action related to toxic activities have not been completely clarified, requiring future studies.

With regard to the results obtained through phytochemical screening, it was possible to identify the presence of flavonoids. To evaluate the presence of flavonoids, four reactions were performed and all of them showed positive results. Shinoda’s reaction slowly revealed the color development that ranged from rosy to reddish; by using the reaction with ferric chloride it was observed the appearance of a color that varied between

![Figure 2: Results of the reactions for identification of flavonoids](image-url)
shades of dark green and brown; for the reaction with sodium hydroxide it was observed the appearance of yellow color that varied in intensity when shaking the tube; and the reaction with aluminum chloride it was noticed the appearance of fluorescence in the region of the filter paper in which the reactive was applied (Figure 2).

The presence of flavonoids in phytochemical screening is justified by the fact that flavonoids are known as one of the broadest groups of secondary metabolites of plants, being important natural pigments widely found in different parts of plants. The great diversity in the structure of flavonoids occurs due to some subtle modifications in its chemical composition, in its basic structure, and it may be through hydroxylation, methylation, acylation, glycosylation, hydrogenation, malonylations and sulfation. [25,26].

Several activities of the flavonoids in the organism have been observed, such as: vasodilatation, antioxidative potential, anti-inflammatory, antiallergic, antitumor, anti-hepatotoxic, antiulcerogenic and antiplatelet activities, as well as antimicrobial and antiviral actions [25].

Thus, considering that phytochemical screening revealed only flavonoids, it is possible to infer that the antimicrobial activity verified from the tested plant derivatives may be related to this class of secondary metabolites.

The reactions of identification of the alkaloids were negative, because there was no formation of precipitates or turbidity in any of the tubes that received the reactives (Figure 3).

Cardiotonic compounds were tested by three reactions. Two of them presented a positive result and one presented a negative result. The first positive reaction was that of Liebermann-Bouchard, because the appearance of a reddish ring in the contact zone of the two layers was observed; the second positive reaction was that of Keller-Killiani because of the appearance of red-brownish coloration in the zone of contact of the liquids and green coloration in the acetic layer. The only negative reaction was Kedde’s reaction, as the intense red-violet color expected to appear was not evident (Figure 4). Thus, we consider the result...
negative for cardiotonics.

In the evaluation of saponins, the reaction was negative, as there was no persistent foam formation (positivity is confirmed when the foam persists for more than 15 seconds) (Figure 5).

In relation to the tannins, the result was also negative, because no precipitation was observed in any of the three reactions proceeded (Figure 6).

Finally, to verify if the plant derivative had anthraquinones, three tests were performed. No free antraquinones were evident (Figure 7), as there was no pink or red coloration in the lower phase of the test tube during the tests. Likewise, O-Glycosides and C-Glycosides were not identified, since no red color appeared in the ammoniacal layer in order to indicate the presence of anthraquinone glycosides with C-O bond or C-C bond, respectively.

Table 2 compiles literature information obtained from a survey in the Scielo databases, using the words: *Euphorbia tirucalli* L., aveloz, phytochemical screening and antimicrobial potential. We selected studies that presented a similar approach to the present study, regarding the phytochemical screening performed.

The comparison between the results showed that the absence of anthraquinones and cardiotonics (in relation to the studies that carried out this investigation) were the common points. In addition, it is important to point out that the material analyzed in the present study was the only one that did not reveal the presence of alkaloids. This result allows us to consider the studied specimen as a potential source of new antimicrobial agents, since the alkaloids represent a group with known toxic components.

Thus, the results obtained in the present study reinforce the need to carry out continuous phytochemical and antimicrobial evaluations with different specimens of *E. tirucalli* for better characterization of antimicrobial properties.

### Acknowledgement

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